



課程名稱：醫學工程概論

(Introduction to biomedical engineering)

A close-up photograph of a silver stethoscope resting on a medical chart with colorful tabs labeled with numbers and letters (1, 3, 5, 7, 8, M, 4, 6).

# 生物系統與生醫訊號分析

## Biomedical Systems and Biological Signal Analysis

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開課單位：通識教育中心

開課學期：102學年度上學期

上課地點：2202 上課時間：星期一3,  
4



# 學習目標：

- 生物系統簡介
- 週期性訊號簡介
- 線性訊號及非線性訊號簡介
- 生醫訊號分析
- 生醫訊號臨床應用簡介





# 週期性訊號簡介

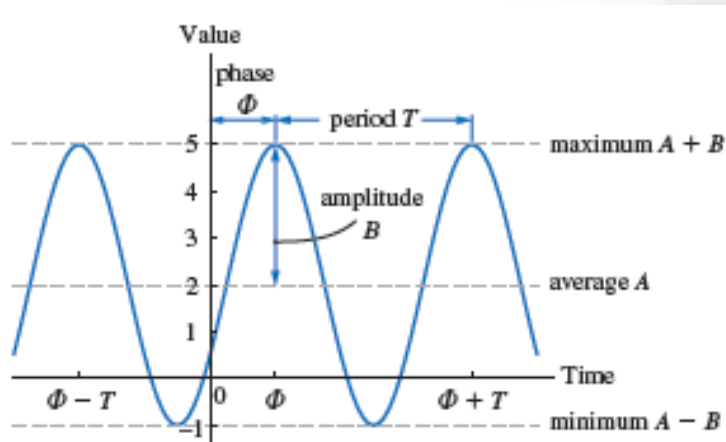
## Sine and Cosine: A Review

In applied mathematics, angles are measured in radians, defined as the distance along the perimeter of the circle of radius 1.

The **basic identity** between radians and degrees is given by  $2\pi \text{ rad} = 360^\circ$ .

## Describing Oscillations with the Cosine

Oscillations that are shaped like the graph of the sine or cosine function are called **sinusoidal**. There are four numbers needed to describe an oscillation with the cosine function: the **average**, the amplitude, the period, and the phase.



- The **amplitude** is the difference between the maximum and the average.
- The **period** is the time between successive peaks.
- The **phase** is the time of the first peak.

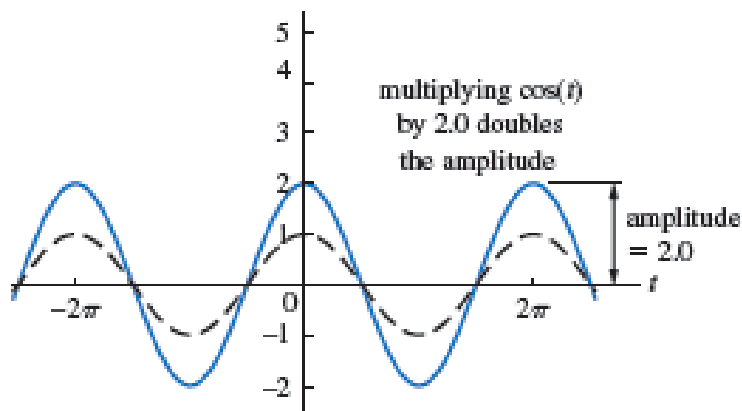


# Building an Oscillation by Shifting and Scaling the Cosine Function



$$f(t) = 2.0 \cos(t).$$

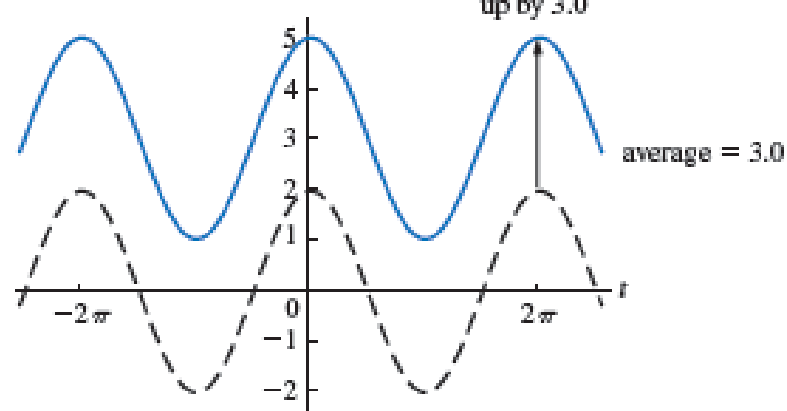
$$3.0 + 2.0 \cos(t)$$



$$f(t) = 3.0 + 2.0 \cos(t)$$

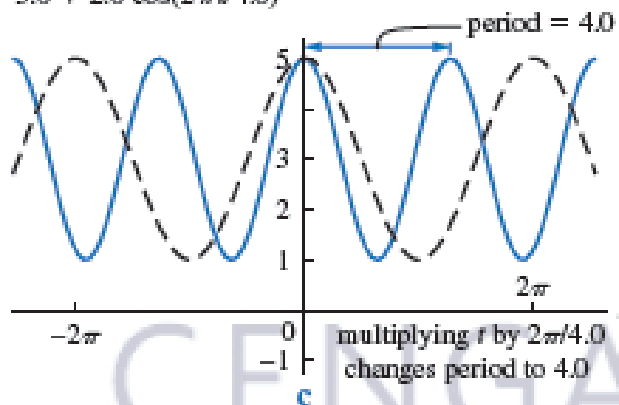
$$3.0 + \cos(t)$$

adding 3.0  
moves graph  
up by 3.0



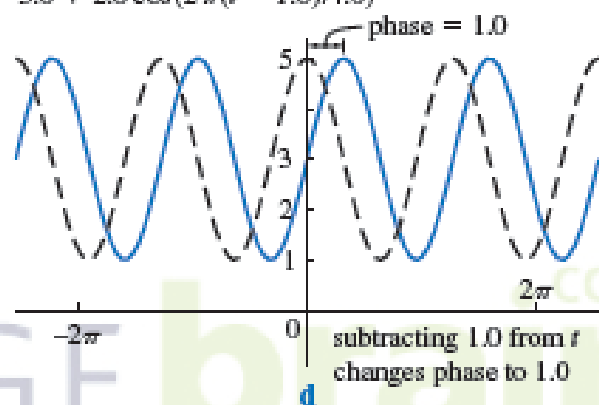
$$f(t) = 3.0 + 2.0 \cos(2\pi t/4)$$

$$3.0 + 2.0 \cos(2\pi t/4.0)$$



$$f(t) = 3.0 + 2.0 \cos(2\pi(t-1)/4)$$

$$3.0 + 2.0 \cos(2\pi(t-1.0)/4.0)$$



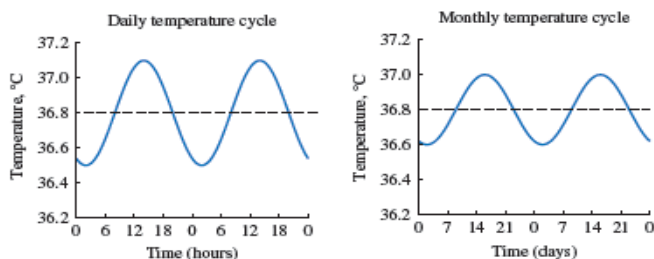


# The Daily and Monthly Temperature Cycles



Women have two cycles affecting body temperature: a daily and a monthly rhythm. The key facts about these two cycles are given in the following table:

|               | Minimum | Maximum | Average | Time of Maximum | Period   |
|---------------|---------|---------|---------|-----------------|----------|
| Daily cycle   | 36.5    | 37.1    | 36.8    | 2:00 p.m.       | 24 hours |
| Monthly cycle | 36.6    | 37.0    | 36.8    | Day 16          | 28 days  |



The amplitude of a cycle is : amplitude = maximum – average.

For the daily cycle, the amplitude is

$$\text{daily cycle amplitude} = 37.1 - 36.8 = 0.3.$$

For the monthly cycle, the amplitude is

$$\text{monthly cycle amplitude} = 37.0 - 36.8 = 0.2.$$

$$P_d(t) = 36.8 + 0.3 \cos\left(\frac{2\pi}{24}(t - 14)\right). \quad P_m(t) = 36.8 + 0.2 \cos\left(\frac{2\pi}{28}(t - 16)\right).$$





# 線性及非線性訊號簡介

We now derive a model of two competing bacterial populations that leads naturally to a discrete-time dynamical system that is not linear. Nonlinear dynamical systems may have more than one equilibrium. By comparing the two equilibria, we will catch a glimpse of an important theme, the stability of equilibria.

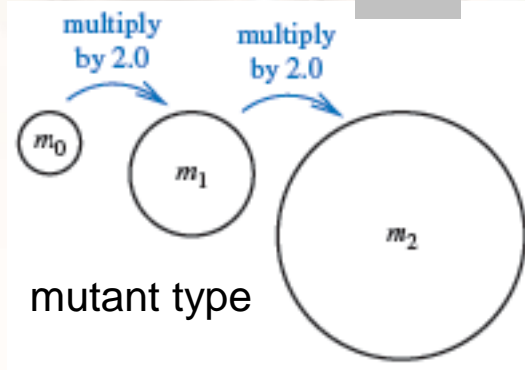
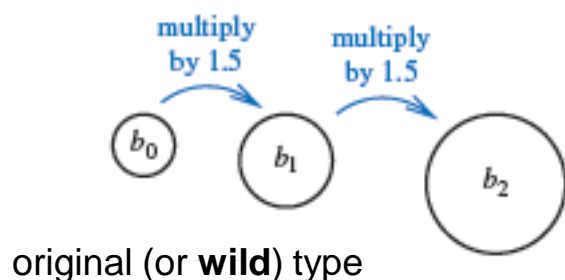
**Model of Selection: An invasion by mutant bacteria**

If the original (or **wild**) type has a per capita reproduction of 1.5 and the mutant type has a per capita reproduction of 2.0, the two populations will follow the discrete-time dynamical systems

$b_{t+1} = 1.5b_t$  discrete-time dynamical system for wild type

$m_{t+1} = 2.0m_t$  discrete-time dynamical system for mutants.

**Selection** occurs when the frequency of a gene (the mutation) changes over time.



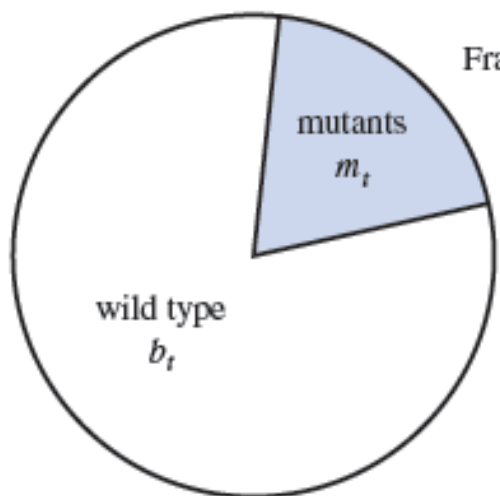
$$m_{t+1} = sm_t$$
$$b_{t+1} = rb_t.$$



# 線性及非線性訊號簡介

A new variable called  $p_t$  was defined to be the fraction of mutants at time  $t$ . Then

$$\begin{aligned} p_t &= \frac{\text{number of mutants}}{\text{total number}} \\ &= \frac{\text{number of mutants}}{\text{number of mutants} + \text{number of wild type}} \\ &= \frac{m_t}{m_t + b_t}. \end{aligned}$$



$$\text{Fraction of mutants} = p_t = \frac{m_t}{m_t + b_t}$$

$$\text{Fraction of wild type} = 1 - p_t = \frac{b_t}{m_t + b_t}$$

$$\begin{aligned} p_{t+1} &= \frac{m_{t+1}}{m_{t+1} + b_{t+1}} \\ &= \frac{sm_t}{sm_t + rb_t} \\ &= \frac{s \frac{m_t}{m_t + b_t}}{s \frac{m_t}{m_t + b_t} + r \frac{b_t}{m_t + b_t}} \\ &= \frac{sp_t}{sp_t + r(1 - p_t)}. \end{aligned}$$



# 線性及非線性訊號簡介

## Example Finding the Updated Fraction

If  $m_t = 2.0 \times 10^5$  and  $b_t = 3.0 \times 10^6$  (Example 1.10.1), the updated populations are

$$m_{t+1} = 2.0m_t = 4.0 \times 10^5$$

$$b_{t+1} = 1.5b_t = 4.5 \times 10^6.$$

The updated fraction of the mutant type,  $p_{t+1}$ , is

$$p_{t+1} = \frac{4.0 \times 10^5}{4.0 \times 10^5 + 4.5 \times 10^6} = 0.0816.$$

We can follow these same steps to find the discrete-time dynamical system for  $p_t$ .  
By definition

$$p_{t+1} = \frac{m_{t+1}}{m_{t+1} + b_{t+1}}.$$

Using the discrete-time dynamical systems for the two types (Equation 1.10.1), we find

$$p_{t+1} = \frac{2.0m_t}{2.0m_t + 1.5b_t}.$$

$$p_{t+1} = \frac{2.0 \frac{m_t}{m_t + b_t}}{2.0 \frac{m_t}{m_t + b_t} + 1.5 \frac{b_t}{m_t + b_t}}.$$

$$p_{t+1} = \frac{2.0p_t}{2.0p_t + 1.5(1 - p_t)}.$$

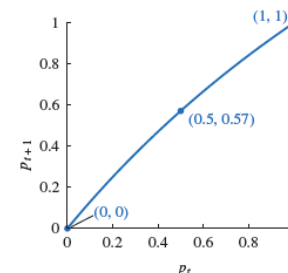




# 線性及非線性訊號簡介

If  $p_t = 0.0625$ , the discrete-time dynamical system tells us that

$$p_{t+1} = \frac{2.0 \cdot 0.0625}{2.0 \cdot 0.0625 + 1.5(1 - 0.0625)} = 0.0816.$$



The discrete-time dynamical system for the fraction is not linear because it involves division. The graph of the function is curved. We plotted it by substituting in representative values for a fraction, which must lie between 0 and 1.

The equilibria of this discrete-time dynamical system are found by solving

$$p^* = \frac{2.0p^*}{2.0p^* + 1.5(1 - p^*)}$$

the equation for the equilibrium

$$p^*(2.0p^* + 1.5(1 - p^*)) = 2.0p^*$$

multiply through by denominator

$$p^*(2.0p^* + 1.5(1 - p^*)) - 2.0p^* = 0$$

move everything to one side

$$p^*(2.0p^* + 1.5(1 - p^*) - 2.0) = 0$$

factor out  $p^*$

$$p^*(2.0p^* + 1.5 - 1.5p^* - 2.0) = 0$$

multiply out terms in parentheses

$$p^*(0.5p^* - 0.5) = 0$$

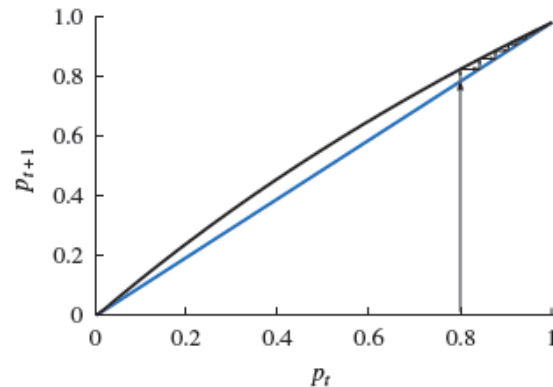
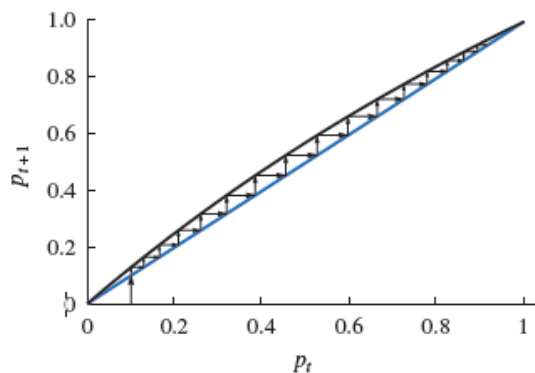
simplify

Extinction of the mutant (at  $p^* = 0$ ), extinction of the wild type (at  $p^* = 1$ )

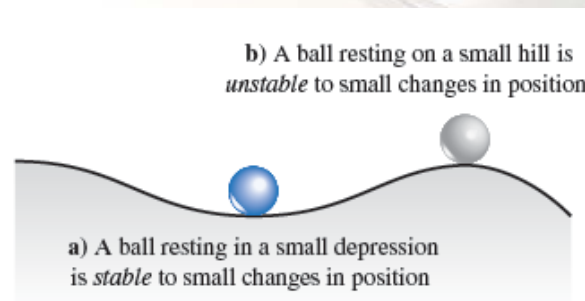


# Stable and Unstable Equilibria

If we started **exactly** at  $p_0 = 0$ , the solution would remain at  $pt = 0$  for all times  $t$ . Similarly, if we started **exactly** at  $p_t = 1$ , the solution would remain at  $pt = 1$  for all times  $t$ . The two equilibria behave quite differently, however, if our starting point is nearby. A solution starting **near**  $p_0 = 0$  moves steadily **away** from the equilibrium. A solution starting **near**  $p_0 = 1$  moves **toward** the equilibrium.



Stable and unstable resting points for a ball



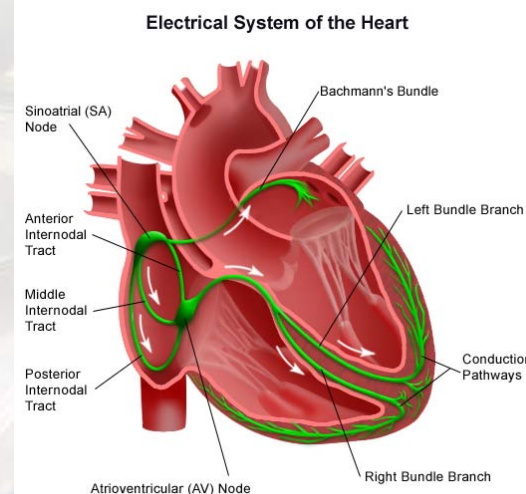
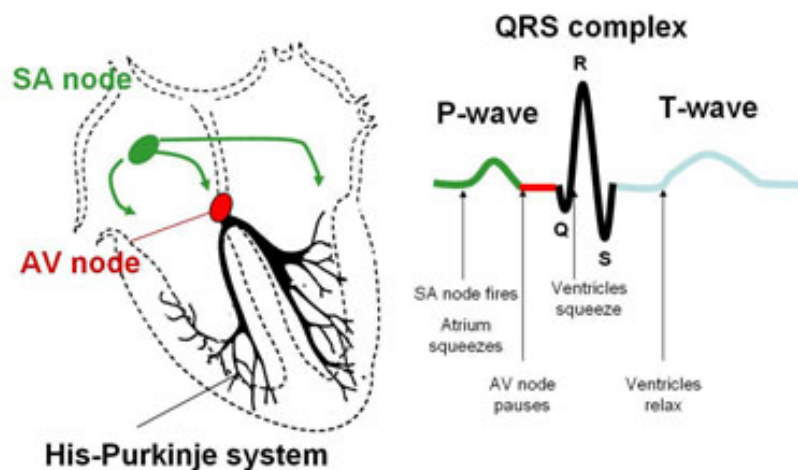


# 生醫訊號臨床應用簡介



## An Excitable Systems: The Heart

- The sinoatrial node (SA node) is the pacemaker, sending regular signals to the atrioventricular node (AV node). The AV node then tells the heart to beat if conditions are suitable.
- Our goal is to understand how simple changes in the parameters of a heart can produce heartbeat patterns called **second-degree block**. With these syndromes, people's hearts either beat half as often as they should or beat normally for a while, skip a beat, and return to beating.





# 生醫訊號臨床應用簡介

## Membrane Potential

Membrane potentials in cells are determined primarily by three factors:

- 1) the concentration of ions on the inside and outside of the cell;
- 2) the permeability of the cell membrane to those ions through specific ion channels;
- 3) by the activity of electrogenic pumps (e.g.  $\text{Na}^+/\text{K}^+ \text{--ATPase}$  and  $\text{Ca}^{++}$  transport pumps) that maintain the ion concentrations across the membrane.

### Equilibrium potential for $\text{K}^+$ ( $E_K$ ; Nernst potential)

$$E_K = \frac{-61}{z} \log \frac{[\text{K}^+]_i}{[\text{K}^+]_o} = -96 \text{ mV}$$

(where  $[\text{K}^+]_i = 150 \text{ mM}$  and  $[\text{K}^+]_o = 4 \text{ mM}$ ; and  $z=1$  because  $\text{K}^+$  is monovalent)

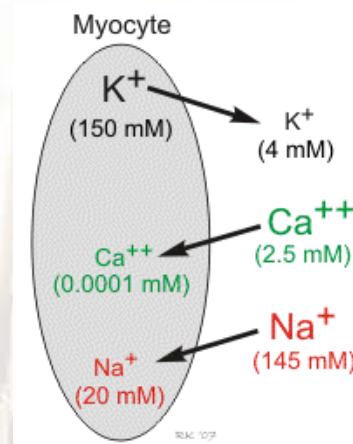
### Equilibrium potential for $\text{Na}^+$ ( $E_{\text{Na}}$ )

$$E_{\text{Na}} = \frac{-61}{z} \log \frac{[\text{Na}^+]_i}{[\text{Na}^+]_o} = +52 \text{ mV}$$

(where  $[\text{Na}^+]_i = 20 \text{ mM}$  and  $[\text{Na}^+]_o = 145 \text{ mM}$ ; and  $z=1$  because  $\text{Na}^+$  is monovalent)

### Equilibrium potential for $\text{Ca}^{++}$ ( $E_{\text{Ca}}$ )

$$E_{\text{Ca}} = \frac{-61}{z} \log \frac{[\text{Ca}^{++}]_i}{[\text{Ca}^{++}]_o} = +134 \text{ mV}$$



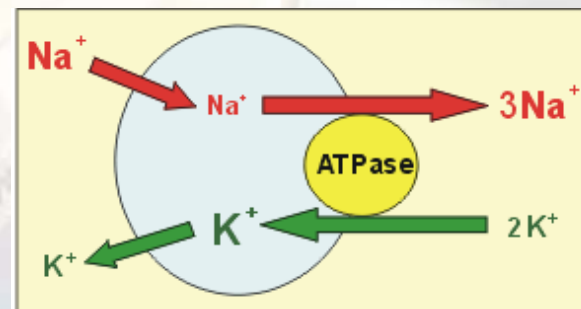


# 生醫訊號臨床應用簡介



## Membrane Potential

- To maintain the concentration gradients for  $\text{Na}^+$  and  $\text{K}^+$ , it is necessary to transport  $\text{Na}^+$  out of the cell and  $\text{K}^+$  back into the cell. There is located on the sarcolemma an energy dependent (ATP-dependent) pump system  **$\text{Na}^+/\text{K}^+$ -ATPase** that that performs this function.
- Pump is electrogenic in nature because it extrudes 3  $\text{Na}^+$  for every 2  $\text{K}^+$  entering the cell.
- Inhibition of this pump, therefore, causes depolarization resulting not only from changes in  $\text{Na}^+$  and  $\text{K}^+$  concentration gradients, but also from the loss of an electrogenic component of the membrane potential.
- By pumping more positive charges out of the cell than into the cell, the pump activity creates a negative potential within the cell.
- Small increases in external  $\text{K}^+$  can stimulate the pump activity and thereby cause hyperpolarization.







# 生醫訊號臨床應用簡介

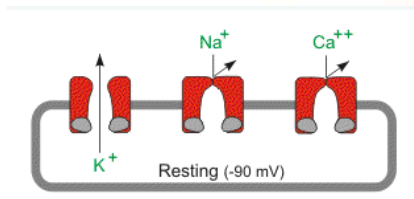


## Action Potentials

Many cells in the body have the ability to undergo a transient repolarization and depolarization that is either triggered by external mechanisms.

**Resting membrane potential (phase 4, about -90 mV)**

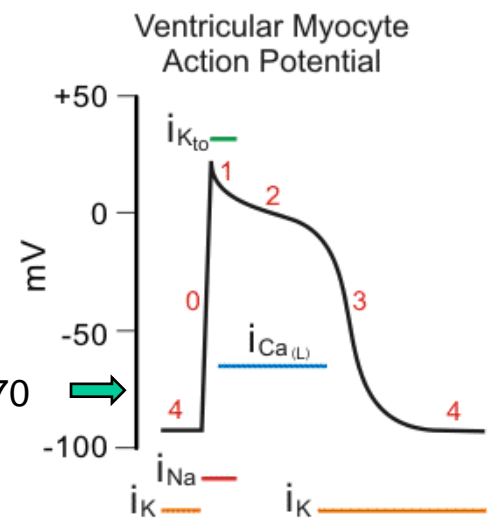
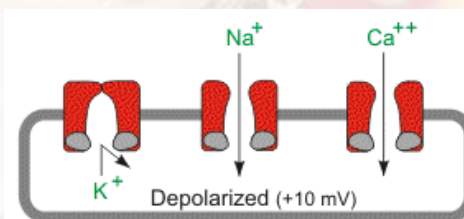
Positive potassium ions are leaving the cell and making the membrane potential more negative inside. At the same time, fast sodium channel and slow calcium channels are closed.



**threshold voltage -70**

**When depolarized to a threshold voltage (phase 0, +10 mV)**

Caused by a transient increase in fast  $\text{Na}^+$  currents ( $I_{\text{Na}}$ ), due to the opening of sodium channels, and concomitant outward directed  $\text{K}^+$  currents as the potassium channel closed.



<http://www.cvphysiology.com/Arrhythmias/A010.htm>





# 生醫訊號臨床應用簡介



## Action Potentials

### Repolarization (phase 1)

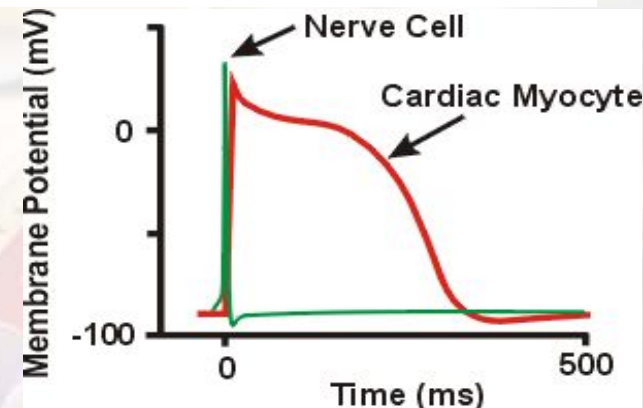
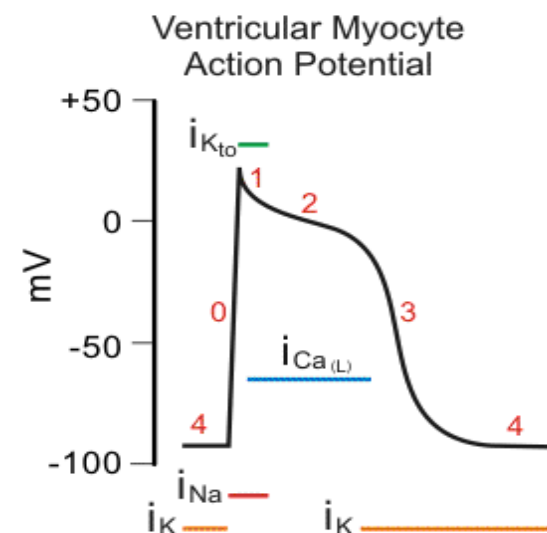
represents an initial repolarization that is caused by the opening of a special type of transient outward  $K^+$  channel ( $K_{to}$ ), which causes a short-lived, hyperpolarizing outward  $K^+$  current ( $I_{Kto}$ ).

### Repolarization (phase 2)

Large increase in slow inward  $gCa^{++}$  occurring at the same time and the transient nature of  $I_{Kto}$ , the repolarization is delayed and there is a plateau phase in the action potential.

### Repolarization (phase 3)

Occurs when  $gK^+$  (and therefore  $I_K$ ) increases, along with the inactivation of  $Ca^{++}$  channels (decreased  $gCa^{++}$ ).



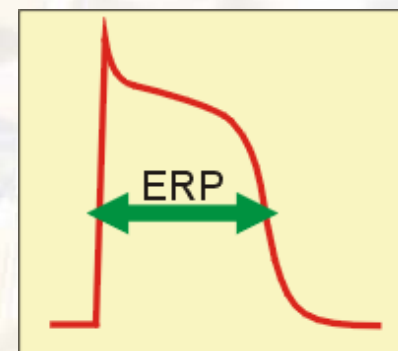
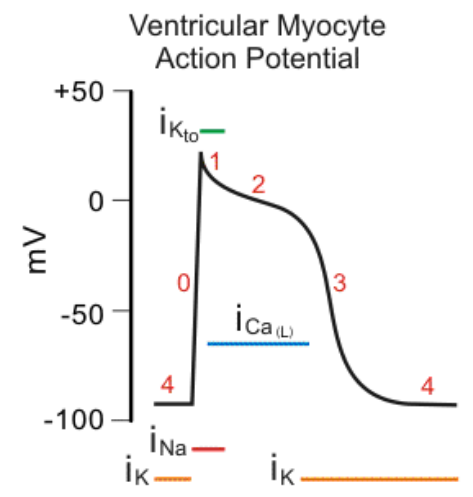


# 生醫訊號臨床應用簡介



## Effective Refractory Period

- Once an action potential is initiated, there is a period of time comprising **phases 0, 1, 2, and part of phase 3** that a new action potential cannot be initiated. This is termed the **effective refractory period (ERP)** or the **absolute refractory period (ARP)** of the cell.
- During the ERP, stimulation of the cell by an adjacent cell undergoing depolarization does not produce new, propagated action potentials. The ERP acts as a **protective mechanism** in the heart by preventing multiple, compounded action potentials from occurring (i.e., it limits the frequency of depolarization and therefore heart rate).





# 生醫訊號臨床應用簡介



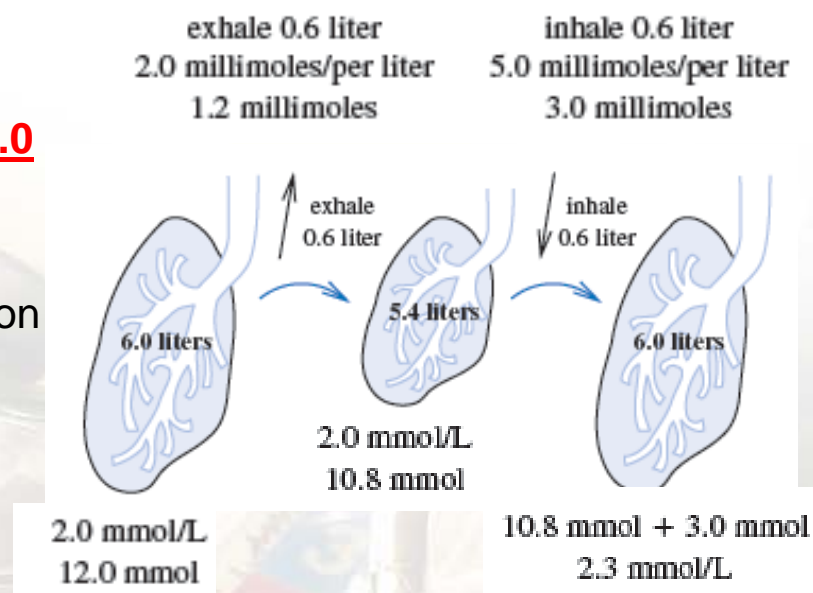
## A Model of Gas Exchange in the Lung

- An adult male lung has a volume of about **6.0 L** when full. With each breath, **0.6 L** of the air in the lungs are exhaled, and replaced by 0.6 L of outside (or **ambient**) air.
- Suppose further that the lung contains a particular chemical with a concentration of **2.0 mmol/L** before exhaling that the lungs contain.
- The ambient air has a chemical concentration of **5.0 mmol/L**. What is the chemical concentration after one breath?

$$\text{concentration} = \frac{\text{total amount}}{\text{volume}}$$

total amount = concentration X volume.

Assumption: Neither air nor chemical is produced or used while breathing.



( A millimole is  $6.02 \times 10^{20}$  molecules)



# 生醫訊號臨床應用簡介

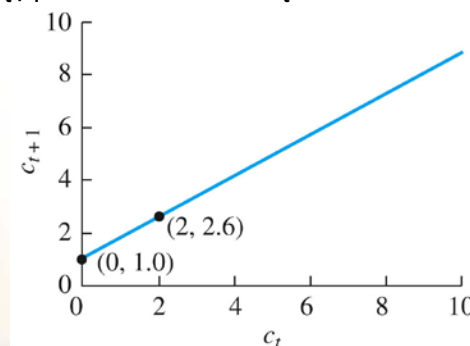


Breathing creates a discrete-time dynamical system. The original concentration of 2.0 mmol/L is updated to 2.3 mmol/L after a breath.

| Step                          | Volume (L) | Total Chemical Concentration (mmol) | Concentration (mmol/L) | What We Did   |
|-------------------------------|------------|-------------------------------------|------------------------|---|
| Air in lungs before breath    | 6.0        | $6.0c_t$                            | $c_t$                  | Multiplied volume of lungs (6.0) by concentration ( $c_t$ ) to get $6.0c_t$ .                                       |
| Air exhaled                   | 0.6        | $0.6c_t$                            | $c_t$                  | Multiplied volume exhaled (0.6) by concentration ( $c_t$ ) to get $0.6c_t$ .  |
| Air in lungs after exhalation | 5.4        | $5.4c_t$                            | $c_t$                  | Multiplied volume remaining (5.4) by concentration ( $c_t$ ) to get $5.4c_t$ .                                      |
| Air inhaled                   | 0.6        | 3.0                                 | 5.0                    | Multiplied volume inhaled (0.6) by ambient concentration (5.0) to get 3.0.  |
| Air in lungs after breath     | 6.0        | $3.0 + 5.4c_t$                      | $0.5 + 0.9c_t$         | Added inhaled chemical (3.0) to remaining chemical ( $5.4c_t$ ) and divided by volume (6.0) to get $0.5 + 0.9c_t$ . |

Discrete-time dynamical system

$$c_{t+1} = 0.5 + 0.9c_t$$

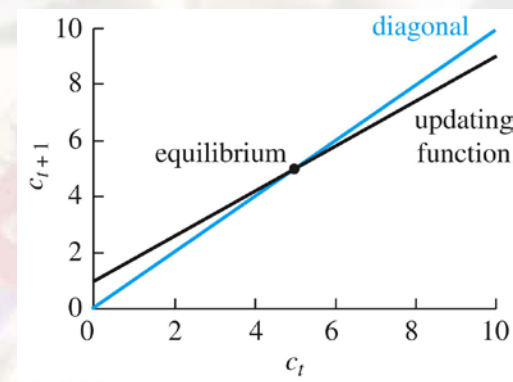


The solutions of the updating equation are equilibria.

$$\begin{aligned}
 c^* &= 0.5 + 0.9c^* && \text{the original equation} \\
 c^* - 0.9c^* &= 0.5 && \text{subtract } 0.9c^* \text{ to get unknowns on one side} \\
 0.1c^* &= 0.5 && \text{do the subtraction} \\
 c^* &= \frac{0.5}{0.1} = 5.0 && \text{divide by 0.1}
 \end{aligned}$$

The equilibrium value is 5.0 mmol/L.

$$c_{t+1} = 0.5 + 0.9 \times 5.0 = 5.0 = c_t$$





# 生醫訊號臨床應用簡介



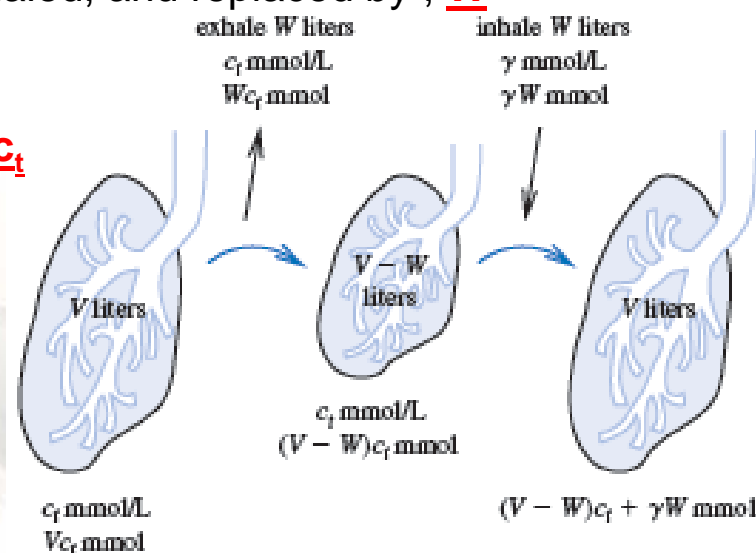
## A “General” Model of Gas Exchange in the Lung

- An adult male lung has a volume of about  $V$  liter when full. With each breath,  $W$  liter of the air in the lungs are exhaled, and replaced by ,  $W$  liter of outside (or **ambient**) air.
- Suppose further that the lung contains a particular chemical with a concentration of  $c_t$  mmol/L before exhaling that the lungs contain.
- The ambient concentration of chemical is  $\gamma$  mmol/L (“gamma”).

Discrete-time dynamical system

$$\begin{aligned} c_{t+1} &= \frac{c_t(V - W) + \gamma W}{V} \\ &= \frac{c_t V - c_t W + \gamma W}{V} \\ &= c_t - c_t \frac{W}{V} + \gamma \frac{W}{V}. \end{aligned}$$

$$q = \frac{W}{V} = \text{fraction of air exchanged}$$



General lung discrete-time dynamical system

$$c_{t+1} = (1 - q)c_t + q\gamma.$$

The term  $c_{t+1}$  is a **weighted average** of the old concentration  $c_t$  and the ambient concentration  $\gamma$ .

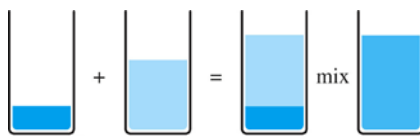




# 生醫訊號臨床應用簡介



**Definition** A weighted average of two values  $x$  and  $y$  is a sum of the form  $qx + (1 - q)y$  for some value of  $q$  between 0 and 1.



Suppose 1.0 L of liquid with a concentration of 10.0 mmol/L of salt are mixed with 3.0 L of liquid with a concentration of 5.0 mmol/L of salt. What is the concentration of the resulting mixture?

$$0.25 \times 10.0 \text{ mmol/L} + 0.75 \times 5.0 \text{ mmol/L} = 6.25 \text{ mmol/L}.$$

Suppose 1.0 L of liquid with a concentration of 10.0 mmol/L of salt are mixed with 3.0 L of liquid with a concentration of 5.0 mmol/L of salt and 1.0 L of liquid with a concentration of 2.0 mmol/L of salt. What is the concentration of the resulting mixture?

$$0.20 \times 10.0 \text{ mmol/L} + 0.60 \times 5.0 \text{ mmol/L} + 0.20 \times 2.0 \text{ mmol/L} = 5.4 \text{ mmol/L}.$$



(1L/5L)



(3L/5L)



(1L/5L)





# 生醫訊號臨床應用簡介

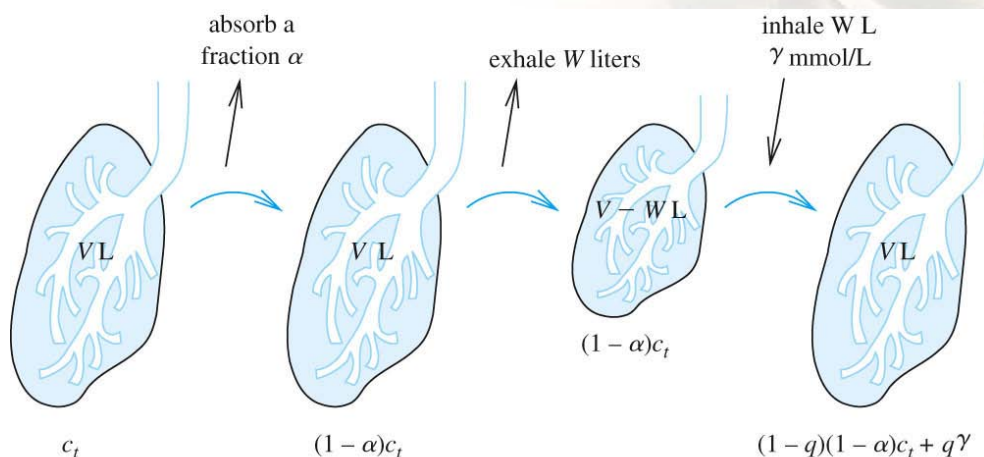


## A “Absorption” Model of Gas Exchange in the Lung

- An adult male lung has a volume of about V liter when full. With each breath, W liter of the air in the lungs are exhaled, and replaced by , W liter of outside (or **ambient**) air.
- Suppose further that the lung contains a particular chemical with a concentration of  $c_t$  mmol/L before exhaling that the lungs contain.
- The ambient concentration of chemical is  $\gamma$  mmol/L (“gamma”).
- Suppose that a fraction  $q$  of air is exchanged each breath, and that a fraction  $\alpha$  of chemical is absorbed

### Lung discrete-time dynamical system

$$c_{t+1} = (1 - q)(1 - \alpha)c_t + q\gamma.$$



$$q = \frac{W}{V} = \text{fraction of air exchanged}$$



# 生醫訊號臨床應用簡介



## Example: Absorption of Oxygen by the Lung

Consider again a lungs that has a volume of 6.0 L(V) and that replaces 0.6 L(W) each breath with ambient air ( $q = 0.6/6$ ). Suppose that we are tracking oxygen, with an ambient concentration of 21% ( $\gamma$ ), and assume that 30% ( $\alpha$ ) of the oxygen in the lungs is absorbed each breath.

The discrete-time dynamical system is then

$$c_{t+1} = 0.9 \times 0.7c_t + 0.1 \times 0.21 = 0.63c_t + 0.021.$$

The equilibrium concentration in the lungs then becomes

$$c^* = 0.63c^* + 0.021$$

$$0.37c^* = 0.021$$

$$c^* = 0.057.$$

The equilibrium concentration of oxygen in the lungs, which is equal to the concentration of oxygen in the air breathed out, would be about 5.7%, or roughly one fourth of the ambient concentration.

$$\begin{aligned} c^* &= (1-q)(1-\alpha)c^* + q\gamma \\ c^* - (1-q)(1-\alpha)c^* &= q\gamma \\ c^*(1 - (1-q)(1-\alpha)) &= q\gamma \\ c^* &= \frac{q\gamma}{1 - (1-q)(1-\alpha)}. \end{aligned}$$

$$\text{Total absorbed per breath} = \alpha c^* V$$